

(19)



Europäisches Patentamt

European Patent Office

Office européen des brevets



(11)

EP 0 997 507 A1

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication:
03.05.2000 Bulletin 2000/18

(51) Int. Cl.⁷: **C09D 11/10**, C09D 11/00,
C09D 171/02, C08G 65/333

(21) Application number: **99202795.3**

(22) Date of filing: **31.08.1999**

(84) Designated Contracting States:
**AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU
MC NL PT SE**
Designated Extension States:
AL LT LV MK RO SI

(30) Priority: **29.10.1998 EP 98203629**

(71) Applicant: **AGFA-GEVAERT N.V.**
2640 Mortsel (BE)

(72) Inventors:
• **Verdonck, Emiel**
2640 Morstel (BE)
• **Vanmaele, Luc**
2640 Morstel (BE)

(54) **New ink compositions for ink jet printing**

(57) An ink composition for ink jet printing contains an amino functionalized polyetheracrylate as radiation curable compound.

EP 0 997 507 A1

Description

FIELD OF THE INVENTION

5 [0001] The present invention relates to a radiation curable ink composition for ink jet printing.

BACKGROUND OF THE INVENTION

10 [0002] In the majority of applications printing proceeds by pressure contact of an ink-loaden printing form with an ink-receiving material which is usually plain paper. The most frequently used impact printing technique is known as lithographic printing based on the selective acceptance of oleophilic ink on a suitable receptor.

[0003] In recent times however so-called non-impact printing systems have replaced classical pressure-contact printing to some extent for specific applications. A survey is given e.g. in the book "Principles of Non Impact Printing" by Jerome L. Johnson (1986), Palatino Press, Irvine, CA 92715, USA.

15 [0004] Among non-impact printing techniques ink jet printing has become a popular technique because of its simplicity, convenience and low cost. Especially in those instances where a limited edition of the printed matter is needed ink jet printing has become a technology of choice. A recent survey on progress and trends in ink jet printing technology is given by Hue P. Le in Journal of Imaging Science and Technology Vol. 42 (1), Jan/Febr 1998, which is hereby included as reference.

20 [0005] In ink jet printing tiny drops of ink fluid are projected directly onto an ink receptor surface without physical contact between the printing device and the receptor. The printing device stores the printing data electronically and controls a mechanism for ejecting the drops image-wise. Printing is accomplished by moving the print head across the paper or vice versa. Early patents on ink jet printers include US 3,739,393, US 3,805,273 and US 3,891,121.

25 [0006] The jetting of the ink droplets can be performed in several different ways. In a first type of process a continuous droplet stream is created by applying a pressure wave pattern. This process is known as continuous ink jet printing. In a first embodiment the droplet stream is divided into droplets that are electrostatically charged, deflected and recollected, and into droplets that remain uncharged, continue their way undeflected, and form the image. Alternatively, the charged deflected stream forms the image and the uncharged undeflected jet is recollected. In this variant of continuous ink jet printing several jets are deflected to a different degree and thus record the image (multideflection system).

30 [0007] According to a second process the ink droplets can be created "on demand" ("DOD" or "drop on demand" method) whereby the printing device ejects the droplets only when they are used in imaging on a receiver thereby avoiding the complexity of drop charging, deflection hardware, and ink recollection. In drop-on-demand the ink droplet can be formed by means of a pressure wave created by the mechanical motion of a piezoelectric transducer (so-called "piezo method"), or by means of discrete thermal pushes (so-called "bubble jet" method, or "thermal jet" method).

35 [0008] Ink compositions for ink jet typically include following ingredients : dyes or pigments, water and/or organic solvents, humectants such as glycols, detergents, thickeners, polymeric binders, preservatives, etc.. It will be readily understood that the optimal composition of such an ink is dependent on the ink jetting method used and on the nature of the substrate to be printed. The ink compositions can be roughly divided in :

- 40
- water based ; the drying mechanism involves absorption, penetration and evaporation;
 - oil based ; the drying involves absorption and penetration;
 - solvent based ; the drying mechanism involves primarily evaporation;
 - hot melt or phase change : the ink vehicle is liquid at the ejection temperature but solid at room temperature ; drying
 - 45 is replaced by solidification;
 - UV-curable ; drying is replaced by polymerization.

[0009] It will be readily understood that the first two types of ink compositions require a receiving medium that is more or less absorptive. On the contrary, for non-absorbent substrates solvent based inks, hot melt inks or UV-curable inks will be better suited.

50 [0010] Early patents on water-based inks include US 3,903,034, US 3,889,269, US 3,870,528, US 3,846,141, US 3,776,742 and US 3,705,043. However, it was recognized early that systems based on water-based inks suffer from a number of disadvantages such as : (a) they require water evaporation and therefore an extensive drying system, especially when printing speed is important; (b) large printed areas tend to cockle, (c) the images are sensitive to wet and dry rubbing, (d) inks of low viscosity tend to tip dry on the orifice which can be avoided by the use of humectants, usually glycols, which then increase viscosity. The use of polar solvent based inks can overcome some of the problems inherent to water-based inks, but in its turn causes other problems such as the possible generation of toxic or inflammable vapours. Therefore efforts were conducted to the development of low-solvent ink compositions. In this research the

concept of UV-curable ink compositions was generated, of which a survey is given hereinafter.

[0011] An important basic patent on ink compositions for ink jet, satisfying the need for a low solvent content, and containing a UV-curable compound is US 4,303,924. It describes an ink jet printing process using charged droplets wherein the ink composition contains (a) a multifunctional unsaturated UV-curable compound, (b) a monofunctional unsaturated compound, (c) a reactive synergist, (d) a colorant, (e) an oil soluble salt for conductivity, (f) a photoinitiator, and (g) an organic polar solvent, preferably in a small amount. Several examples of monomers containing acrylate, epoxy, and vinyl functional groups are disclosed.

[0012] In EP 0 071 345 a jet ink composition is claimed comprising (A) a cationically polymerizable epoxy resin chosen from particular classes, (B) a photoinitiator, (C) a colorant, (D) a blend of organic solvents.

[0013] In US 4,680,368 a UV-curable ink, not limited to ink jet, is disclosed comprising (A) a poly(urethane-(meth)acrylate), (B) a radically polymerizable compound and (C) a photoinitiator.

[0014] According to US 4,978,969 the ink composition comprises 12-80 % of a UV curable adhesive, 3-10 % of a pigment, and 10-40 % of a solvent.

[0015] In EP 0 456 039 B1 an ink composition for ink jet is disclosed that is free of volatile organic solvent and contains a colorant, a polar conductive compound, and one or more monomers. In the analogous EP 0 540 203 B1 a non-conductive ink composition, free of volatile solvent, is disclosed, said composition again comprising one or more monomers and a colorant.

[0016] In US 5,270,368 the ink composition contains at least two acrylate types, being an aromatic acrylate with carboxyl groups, and an epoxy acrylate.

[0017] According to EP 0 658 607 an aqueous ink contains a pigment, a water-soluble resin for dispersing the pigment, a water-soluble UV-curable monomer and a photoinitiator.

[0018] In US 5,623,001 an ink is described comprising (a) 20-75 % water, (b) a water-mixable UV-curable compound, preferably an acrylate oligomer, (c) a photoinitiator and (d) a colorant.

[0019] According to US 5,641,346 the ink jet ink contains a colorant, a liquid phase comprising water, and an epoxy compound and/or a vinyl ether compound.

[0020] In WO 97/31071 a radiation-curable ink jet composition is described comprising from 80% to 95% of a polyfunctional (poly)alkoxylated acrylate monomer.

[0021] Summarizing, a radiation curable ink composition may in general contain one or more radiation curable prepolymers, or oligomers, radiation curable monomers or reactive diluents, optionally one or more photoinitiators, colorants, and other additives.

[0022] Many classes of radiation curable compounds are not suitable in radiation curable compositions for ink jet printing due to their high viscosities. On the other hand, using other classes of radiation curable compounds with lower viscosities results in a lower polymerization speed.

[0023] The present invention extends the teachings on radiation curable ink compositions for ink jet.

OBJECTS OF THE INVENTION

[0024] It is an object of the present invention to provide an ink composition for ink jet printing which shows good curing speed.

[0025] It is a further object of the present invention to provide an ink composition for ink jet printing which overcomes the disadvantage of high viscosity.

SUMMARY OF THE INVENTION

[0026] The above mentioned objects are realised by providing a radiation curable ink composition for ink jet printing comprising as radiation polymerizable compound an amino functionalized polyetheracrylate.

DETAILED DESCRIPTION OF THE INVENTION

[0027] Representative examples of commercially available compounds corresponding to the amino functionalized polyetheracrylate type used in accordance with the present invention are listed in the following table 1 without being limited thereto.

TABLE 1

Laromer LR 8869 (ex BASF)
Laromer LR 8889 (")

TABLE 1 (continued)

Actilane 584 (ex Akcros Chemicals)
 Actilane 587 (")
 5 Actilane 595 (")
 Craynor CN 501 (ex Cray Valley)
 Craynor CN 502 (")
 10 Craynor CN 550 (")
 Craynor CN 551 (")
 Crodamer UVA-420 (ex Croda Resins)
 Crodamer UVA-430 ("

15 [0028] The molecular weight of these compounds may vary between 300 and 10000.

[0029] The most preferred compound is Craynor CN 501, an amino functionalized polyetheracrylate derived from trimethylolpropanetriacrylate.

20 [0030] We will now describe systematically the principal other ingredients of the ink composition according to the invention.

Initiators

25 [0031] In a preferred embodiment, the initiator is a photoinitiator. The photoinitiators can be divided in compounds that are suited for cationic polymerization and compounds suited for free radical polymerization.

[0032] References on photoinitiators include following : P.K.T. Oldring (ed.), "Chemistry and Technology of UV and EB Formulation for Coatings, Inks and Paints: Vol. 3 Photoinitiators for Free Radical and Cationic Polymerization," SITA Technology Ltd., London, UK (1991); N.S. Allen, "Photoinitiators for UV and visible curing of coatings: mechanism and properties", *Journal of Photochemistry and Photobiology, A: Chemistry* 100 (1996) 101-107; J.V. Koleske, "A radiation-cure primer", *Journal of Coatings Technology*, Vol. 69, No. 866, March 1997, 29-38.

30 [0033] Disclosures specific on photoinitiators for cationic polymerisation include : J.V. Crivello, "The Chemistry of Photoacid Generating Compounds", Proceedings of the ACS Division of Polymeric Materials: *Science and Engineering*, Vol. 61, pages 62-66, (1989); J.V. Crivello and J.H.W. Lam, "Complex Triarylsulfonium Salt Photoinitiators I. The Identification, Characterization, and Synthesis of a New Class of Triarylsulfonium Salt Photoinitiators," *Journal of Polymer Science, Polymer Chemistry Edition*, Vol. 18, 2677-2695 (1980); J.V. Crivello and J.H.W. Lam, "Complex Triarylsulfonium Photoinitiators II. The Preparation of Several New Complex Triarylsulfonium salts and the Influence of Their Structure in Photoinitiated Cationic Polymerization," *Journal of Polymer Science, Polymer Chemistry Edition*, Vol. 18, pages 2697-2714 (1980); J.V. Crivello and J.H.W. Lam, "Diaryliodonium Salts A New Class of Photoinitiators for Cationic Polymerization," *Macromolecules*, Vol. 10, pages 1307-1315 (1977); and J.V. Crivello, J.L. Lee and D.A. Conlon, "Developments in the Design and Applications of Novel Thermal and Photochemical Initiators for Cationic Polymerization", *Makromol. Chem. Macromolecular Symposium*, Vol. 13/14, pages 134-160 (1988).

40 [0034] Particularly preferred are the diaryl iodonium salts and their derivatives, the triaryl sulfonium salts and their derivatives, and the triphenyl phosphonium salts and their derivatives, with examples of alkyl and aryl substituents. Very recently, there have been described new types of photoinitiators for cationic polymerization such as triarylsulphonium-tetrakis(pentafluorophenyl)-borate (RHODORSIL 2074, Rhône-Poulenc Chimie), by C. Priou et al. in the Conference Proceedings of Radtech Europe '97, p. 314, and such as onium salts with specific light absorption characteristics in WO 97/47660 (Nippon Kayaky KK).

45 [0035] Useful photoinitiators for free radical polymerization are e.g. LUCIRIN LR8953 (ex BASF), IRGACURE 819 and 907 (ex Ciba-Geigy), DAROCUR 4865 (ex Ciba-Geigy), and isopropylthioxanthenes, e.g. QUANTACURE ITX (ex Rahn AG). Other useful photoinitiators for free radical polymerization are polymeric, oligomeric, respectively copolymerizable photoinitiators such as discussed by M. Visconti et al. respectively W. Davies et al. in the Conference papers 6, respectively 7, of the Radcure Coatings and Inks, Curing and Performance Conference (22-23 June 1998). Such photoinitiators are e.g. ESACURE KIP150, and ESACURE KT 37 and KT 55 (ex Lamberti), and acrylated IRGACURE 2959 or IRGACURE 2959 modified melamine acrylate (ex Akcros Chemicals).

55 [0036] Additional examples of suitable initiators are disclosed in following patents : U.S. Pat. Nos. 4,683,317, 4,378,277, 4,279,717, 4,480,368, 4,443,495, 4,303,924, 4,751,102, 4,334,970, 5,270,368, 5,395,724, and EP 0 540 203, EP 0 568 607 and EP 0 659 039.

[0037] All disclosures cited are totally incorporated herein by reference.

[0038] Sometimes, it is also desirable to include, as well as a primary photoinitiator, a co-initiator, also called initiator synergist which is preferably of the amine type, e.g. the aminobenzoate type. The latter types of co-initiators are generally being used with the xanthone/thioxanthone types of primary photoinitiator. More examples can be found in the Oldring reference cited above.

[0039] The photoinitiator and occasionally the coinitiator are preferably present in an amount from 0.2 to 20 % by weight and most preferably between 1 and 10 %.

Colorants

[0040] Inks of the present invention preferably contain a colorant. Any colorant may be used to impart the desired color to the ink. In embodiments of the present invention the colorant may include at least one pigment, one dye, or a combination thereof.

[0041] A wide variety of organic and inorganic dyes and pigments, alone or in combination may be selected for use in the ink compositions of this invention. The pigment particles should be sufficiently small to permit free flow of the ink through the ink jet printing device, especially at the ejecting nozzles that usually have a diameter ranging from 10 μm to 50 μm . The pigment particle size also has an influence on the pigment dispersion stability, which is critical throughout the life of the ink. It is also desirable to use small particles for maximum color strength.

[0042] Accordingly, the average particle diameter may be from about 0.005 μm to about 15 μm . Preferably, the pigment particle size may range from about 0.005 to about 5 microns, more preferably from about 0.005 to about 1 μm , and most preferably from about 0.005 to about 0.3 μm . Pigment particle sizes outside these ranges may, of course, be used as long as the objectives of the present invention are achieved.

[0043] Very fine dispersions of pigments and methods for their preparation are disclosed in e.g. EP 0 776 952, US 5,538,548, US 5,443,628, EP 0 259 130, US 5,285,064, EP 0 429 828, and EP 0 526 198.

[0044] The pigment can be black, cyan, magenta, yellow, red, blue, green, brown, mixtures thereof, and the like. For example, suitable pigment materials include carbon blacks such as Regal 400R, Mogul L, Elftex 320 from Cabot Co., or Carbon Black FW18, Special Black 250, Special Black 350, Special Black 550, Printex 25, Printex 35, Printex 55, Printex 150T from Degussa Co., and Pigment Black 7. Additional examples of suitable pigments are disclosed in, for example, U.S. Pat. No. 5,389,133 to Gundlach et al., the entire disclosure of which is incorporated herein by reference.

[0045] Suitable pigments include, for instance, C. I. Pigment Yellow 17, C. I. Pigment Blue 27, C. I. Pigment Red 49:2, C. I. Pigment Red 81:1, C. I. Pigment Red 81:3, C. I. Pigment Red 81:x, C. I. Pigment Yellow 83, C. I. Pigment Red 57:1, C. I. Pigment Red 49:1, C. I. Pigment Violet 23, C. I. Pigment Green 7, C. I. Pigment Blue 61, C. I. Pigment Red 48:1, C. I. Pigment Red 52:1, C. I. Pigment Violet 1, C. I. Pigment White 6, C. I. Pigment Blue 15, C. I. Pigment Yellow 12, C. I. Pigment Blue 56, C. I. Pigment Orange 5, C. I. Pigment Black 7, C. I. Pigment Yellow 14, C. I. Pigment Red 48:2, C. I. Pigment Blue 15:3, C. I. Pigment Yellow 1, C. I. Pigment Yellow 3, C. I. Pigment Yellow 13, C. I. Pigment Orange 16, C. I. Pigment Yellow 55, C. I. Pigment Red 41, C. I. Pigment Orange 34, C. I. Pigment Blue 62, C. I. Pigment Red 22, C. I. Pigment Red 170, C. I. Pigment Red 88, C. I. Pigment Yellow 151, C. I. Pigment Red 184, C. I. Pigment Blue 1:2, C. I. Pigment Red 3, C. I. Pigment Blue 15:1, C. I. Pigment Blue 15:3, C. I. Pigment Blue 15:4, C. I. Pigment Red 23, C. I. Pigment Red 112, C. I. Pigment Yellow 126, C. I. Pigment Red 169, C. I. Pigment Orange 13, C. I. Pigment Red 1-10, 12, C. I. Pigment Blue 1:X, C. I. Pigment Yellow 42, C. I. Pigment Red 101, C. I. Pigment Brown 6, C. I. Pigment Brown 7, C. I. Pigment Brown 7:X, C. I. Pigment Black 11, C. I. Pigment Metal 1, C. I. Pigment Metal 2, C. I. Pigment Yellow 128, C. I. Pigment Yellow 93, C. I. Pigment Yellow 74, C. I. Pigment Yellow 138, C. I. Pigment Yellow 139, C. I. Pigment Yellow 154, C. I. Pigment Yellow 185, C. I. Pigment Yellow 180, C. I. Pigment Red 122, C. I. Pigment Red 184, and bridged aluminum phthalocyanine pigments.

[0046] Furtheron the pigment may be chosen from those disclosed in *Industrial Organic Pigments, Production, Properties, Applications*, second edition, W. Herbst, K. Hunger ; VCH, 1997.

[0047] Most preferred pigments are Pigment Yellow 128, 93, 17, 74, 138, 139, 154, 185, 180; Pigment Red 122, 57:1, 184; Pigment Blue 15:3, 15:4, and carbon black.

[0048] The pigment may, but need not, be in the form of a dispersion comprising a dispersant also called pigment stabilizer. The latter may be, for example, of the polyester, polyurethane or polyacrylate type, especially in the form of a high molecular weight block copolymer, and would typically be incorporated at 2.5% to 100% by weight of the pigment. Suitable examples are DISPERBYK (ex BYK Chemie) or SOLSPERSE (ex Zeneca) dispersants. A detailed list of non-polymeric as well as some polymeric dispersants appears in, for example, McCutcheon's Functional Materials, North American Edition, Manufacturing Confectioner Publishing Co., Glen Rock, N.J., pp. 110-129 (1990), the entire disclosure of which is incorporated herein by reference.

[0049] Other pigment stabilizers are disclosed in DE 19636382, US 5,720,802, US 5,713,993, PCT/GB95/02501, US 5,085,689 and GB 2303376.

[0050] The pigment or dye may be present in the ink composition in any effective amount, generally from about 0.5

to about 20 percent by weight of the ink.

Other monomers, oligomers or reactive diluents usable in combination with the radiation curable compounds of the invention

[0051] A wide variety of photopolymerizable and photocrosslinkable compounds can be used in combination with the amino functionalized polyetheracrylates of the present invention.

[0052] In many cases this second photopolymerizable compound will take part only to a limited extent in the curing reaction but will serve primarily as so-called reactive diluent in order to decrease the viscosity of the final ink composition.

[0053] Suitable monomers include e.g. the monomers disclosed in DE-OS Nos. 4005231, 3516256, 3516257, 3632657 and US 4,629,676, and WO 97/31071.

[0054] The photopolymerizable composition may also comprise polymers, prepolymers and/or oligomers and/or reactive diluents comprising one or more polymerizable functions.

[0055] Suitable prepolymers and reactive diluents for use in radiation curable compositions such as the ink composition of the present invention may be selected from the group consisting of unsaturated urethane (meth) acrylates, epoxy (meth) acrylates, polyolacrylates, polyether(meth)acrylates and polyester(meth)acrylates as described e.g. in "Chemistry & Technology of UV and EB formulation for coatings, inks and paints" Vol.2 : Prepolymers and Reactive diluents for UV and EB curable formulations." Ed. G. WEBSTER - SITA Technology - London (1996).

[0056] A survey of UV-curable coating compositions is given e.g. in the periodical "Coating" 9/88, p. 348-353.

[0057] Other usable prepolymers and oligomers belong to the class of aliphatic and aromatic polyester-urethane acrylates. The structure of polyester-urethane acrylates is given in the booklet "Radiation Cured Coatings" by John R. Constanza, A.P. Silveri and Joseph A. Vona, published by Federation of Societies for Coatings Technology, 1315 Walnut St. Philadelphia, PA 19107 USA (June 1986) p. 9.

[0058] A preferred second monomer used in combination with the radiation curable compound of the present invention is chosen from the classes of mono- or diacrylates, vinyl ethers, and vinyl ether (meth) acrylates.

[0059] It will be clear that all these cited monomers, prepolymers, polymers and oligomers can be used in admixture.

Other additives

[0060] Inks of the present invention may include additives such as biocides, buffering agents, anti-mold agents, pH adjustment agents, electric conductivity adjustment agents, chelating agents, anti-rusting agents, polymerisation inhibitors, light stabilizers, and the like. Such additives may be included in the ink jet inks of the present invention in any effective amount, as desired.

[0061] Examples of pH controlling agents suitable for inks of the present invention include, but are not limited to, acids; bases, including hydroxides of alkali metals such as lithium hydroxide, sodium hydroxide and potassium hydroxide. The amount included will depend, of course, on the specific component being included.

[0062] Furtheron, the ink composition of the present invention may also comprise surfactants and photoinitiator stabilizers. Suitable photoinitiator stabilizers include those disclosed in EP 0 465 039. Suitable surfactants are preferably of the non-ionic type, for example FLUORAD FC430 (ex 3M Corp.). Such surfactants when present are preferably included in an amount of 0.1% to 10% by weight of the total composition.

[0063] Compositions according to the present invention may contain organic solvents, such as alcohols, fluorinated solvents and dipolar aprotic solvents. Preferably methanol, ethanol, propanol, 1-butanol, 1-pentanol, 2-butanol, t.-butanol, glycol, glycolethers, N-methylpyrrolidone, N,N-dimethylacetamid, N, N-dimethylformamid, 2,4-pentanedione, and hexafluoroacetone are used.

[0064] The inks of the present invention may contain a dendrimer.

[0065] Dendrimers are radially symmetrical molecules of a STARBURST.TM. topology comprised of an initiator core, such as nitrogen, ethyleneimine, and the like, interior layers attached to the core and comprised of a suitable number of arms, for instance, two to four arms, each arm being comprised of repeating units with the number of repeating units in each arm being considered the generation of the dendrimer, and terminal groups functionality, such as, for example, a primary amine attached to the outmost generation, which dendrimers are illustrated, for example, in U.S. Pat. Nos. 4,507,466; 4,631,337, 4,558,120, 4,568,737, and 4,587,329, and in Tomalia et al., *Angewandte Chemie, Int. Ed. Engl.* 29, 138 (1990). The size and shape of the STARBURST.TM. dendrimer molecule and the functional groups present in the dendrimer molecule can be controlled by the choice of the initiator core, the number of generations, and the choice of repeating units employed at each generation.

[0066] The choice of the dendrimer components can affect the properties of the dendrimers. The initiator core type can affect the dendrimer shape producing, for example, spheroid-shaped dendrimers, cylindrical- or rod-shaped den-

drimers, or ellipsoid-shaped dendrimers. Sequential building of generations determines the dimensions of the dendrimers and the nature of its interior. Examples of suitable core materials include ammonia, polyfunctional alcohols, such as pentaerythritol or tris-(hydroxymethyl)ethane, 1,1,1-tris-(4'-hydroxyphenyl)ethane, polyfunctional amines, such as ethylene diamine, linear polyethyleneimines, and the like. The chemical functionality of the repeating unit in the interior layers can include, for example, amidoamines, such as aminoethyl acetamide, imines, such as diethylene diimine, or ethers like those obtained from materials such as, for example, 3,5-dihydroxyethyl benzyl alcohol. The terminal functionalities include, for example, amino groups, hydroxyl groups, carboxylic acid groups, carboxylates, esters, amides, phosphates, sulfonates, and the like. The synthesis of dendrimers usually occurs by a divergent approach that involves the initial reaction of a monomer with the initiator core, followed by exhaustive reaction of the resulting functional groups with a difunctional compound, such as a diamine, including, for example, ethylene diamine, to afford the next generation of reactive amino groups. Thus, for example, ethylene diamine can be suitably reacted first with methyl acrylate to produce a compound such as N,N,N',N'-tetra(methoxycarbonyl)ethyl)ethylene diamine. The aforesaid compound can be reacted in the next step with ethylene diamine to produce an amidoamine dendrimer having a generation number of zero, a molecular weight of 517, and four primary amino groups at the surface. Repetition of the above two-step procedure leads to subsequent generations.

[0067] An alternate synthetic route uses a convergent growth synthesis as described in detail in Hawker et al., *J. Amer. Chem. Soc.*, 112, 7638 (1990).

[0068] The dendrimer may have other groups or segments, in addition to amino groups. For instance, the dendrimer may have a dye covalently attached to it, or it may have certain functional groups grafted onto it.

[0069] The dendrimers may be grafted with, for example, alkylene oxide oligomers or polymers, wherein the alkylene has 1-12 carbon atoms and the degree of polymerization of the alkylene oxide is in the range of from about 2 to about 100. The amount of grafting can be in any suitable range, preferably below 50% of the amino groups, and even more preferably below 10% of the amino groups. Grafting of ethylene oxide on the dendrimer can be performed by any suitable means known to those of ordinary skill in the art. For instance, a polyethylene glycol monomethyl ether of suitable molecular weight can be converted to polyethylene glycol monomethyl ether p-toluene sulfonate by suitably reacting with p-toluenesulfonyl chloride and pyridine, and the sulfonate then reacted with the dendrimer under suitable conditions, as known to those of ordinary skill in the art. Grafted dendrimers can also be obtained from Dendritech, Inc. in Midland, Mich.

[0070] Preferred dendrimers for use in the preparation of the ink composition of the present invention include those having terminal amine functionality at the surface. It is further preferred that the dendrimer has a molecular weight in the range from about 300 to about 100,000, a generation number of from 0 to 10, a surface amine group concentration of from about 3 to about 4,100, and a molecular diameter of from about 1 nm to about 1000 nm. More preferred dendrimers are those having terminal primary amine functionality. It is also more preferred that the dendrimer has a molecular weight in the range from about 500 to about 30,000, a generation number of from 0 to about 5, a surface group concentration of from about 4 to about 150, and a molecular diameter of from about 1 nm to about 150 nm. It is also preferred that the polydispersity index (Mw/Mn) of the dendrimer is low, preferably in the range of from about 1.1000 to about 1.0001, and more preferably in the range of from about 1.001 to about 1.0001. Examples of dendrimers prepared by the divergent approach include the STARBURST.TM. dendrimers available from Dendritech, Inc. These dendrimers from Dendritech, Inc. are polyamidoamines (PAMAMs) having primary amine terminal surface functionality, and made of ethylene diamine core and sequenced copolymers of ethylene diamine and methyl acrylate. They have a polydispersity index of 1.0007.

[0071] The dendrimer is present in the ink composition in an amount sufficient to provide sufficient adhesion of the ink components to the printing surface, and also to provide sufficient water resistance and cold and hot humidity resistance. The amount of the dendrimer is preferably in the range of from about 0.1% to about 10% by weight, more preferably in the range of from about 0.5% to about 2% by weight, and even more preferably in the range of from about 1% by weight to about 2% by weight, of the ink composition.

[0072] In formulating the final ink jet ink compositions of the present invention, certain physical properties should be satisfied. For example, ink compositions for use in ink jet recording processes should have appropriate viscosity and surface tension characteristics. In the present invention, it is preferred that the ink jet ink composition has a viscosity of from about 1 to about 75 mPa.s at 25 °C. The surface tension is preferably from 20 to 72 mN/m and most preferably from 20 to 60 mN/m.

[0073] Apparatuses for radiation curing are known to those skilled in the art and are commercially available. For example, the curing proceeds with medium pressure mercury vapour lamps with or without electrodes, or pulsed xenon lamps. These ultraviolet sources usually are equipped with a cooling installation, an installation to remove the produced ozone and optionally a nitrogen inflow to exclude air from the surface of the product to be cured during radiation processing. An intensity of 40 to 240 W/cm in the 200-400 nm region is usually employed. An example of a commercially available ultraviolet medium-pressure electrodeless mercury vapour lamp is the model VPS/I600 curing system of Fusion UV systems Ltd., UK. A pulsed xenon flash lamp is commercially available from IST Strahlentechnik GmbH,

Nürtingen, Germany. Using the Fusion model one has also the possibility to use metal halide doped Hg vapour or XeCl excimer lamps, each with its specific UV emission spectrum. This permits a higher degree of freedom in formulating the curing composition : a more efficient curing is possible using the lamp with the most appropriate spectral characteristics.

[0074] High energy ionizing radiation such as X-rays, gamma rays, beta rays and accelerated electrons can also be used to accomplish curing of the ink composition.

[0075] The ink jet receiver materials to which the ink composition of the present invention can be jetted are not limited and include e.g. paper, coated paper, polyolefin coated paper, cardboard, wood, composite boards, plastic, coated plastic, canvas, textile, metal and ceramics.

[0076] The present invention will now be illustrated by the following examples without however being limited thereto.

EXAMPLES

Example 1

[0077] The general composition of the non-colored radiation curable inks of the invention was:

*Ultraviolet polymerizing amino modified polyetheracrylate of the invention (see above: table 1);
Ultraviolet polymerizing reactive diluent Radcure DPGDA (UCB, dipropyleneglycoldiacrylate);
N-methyl diethanol amine (NMDA) as co-initiator or synergist; ethanol;
isopropylthioxanthone (Quantacure ITX, Rahn AG).*

[0078] Substituting the amino modified polyetheracrylate of the invention by another oligomer gave rise to a comparative ink composition.

[0079] All inks were prepared on a basis of a total final weight of 20 g. All ink compositions are indicated in table 2 in weight percentage. All inks contained 2% wt of NMDA, 2%wt of ethanol and 10% wt of Quantacure ITX.

[0080] Firstly, the Radcure DPGDA was added to the oligomer of the invention or to the comparative oligomer. The resulting mixture was stirred for a couple of minutes until the added diluent was completely dissolved. As a third respectively a fourth ink component the liquid NMDA respectively the ethanol was added while stirring for about five minutes to complete the solution step. As the last compound the solid photoinitiator Quantacure ITX was added. The resulting mixture was stirred for about 1 hour in order to completely dissolve the ITX.

TABLE 2

ink composition of non-colored inks of the invention for free radical polymerization								
Ink number	Oligomer of the invention		Comparative oligomer		wt % Radcure DPGDA	wt % NMDA	wt % ethanol	wt % Quantacure ITX
	name	wt %	name	wt %				
1-1	Craynor CN 501	49.8			33.2	5.0	2.0	10.0
1-2	Craynor CN 501	33.2			49.8	5.0	2.0	10.0
1-3	Laromer LR 8869	49.8			33.2	5.0	2.0	10.0
1-4	Laromer LR 8869	33.2			49.8	5.0	2.0	10.0
1-5	Laromer LR 8889	49.8			33.2	5.0	2.0	10.0
1-6	Laromer LR 8889	33.2			49.8	5.0	2.0	10.0
1-7			Craynor CN435	49.8	33.2	5.0	2.0	10.0

TABLE 2 (continued)

ink composition of non-colored inks of the invention for free radical polymerization								
Ink number	Oligomer of the invention		Comparative oligomer		wt % Rad-cure DPGDA	wt % NMDA	wt % ethanol	wt % Quanta-cure ITX
	name	wt %	name	wt %				
1-8			<i>Craynor CN435</i>	33.2	49.8	5.0	2.0	10.0
1-9			<i>Laromer LR 8765</i>	49.8	33.2	5.0	2.0	10.0
1-10			<i>Laromer LR 8765</i>	33.2	49.8	5.0	2.0	10.0
1-11			<i>Laromer LR 8986</i>	49.8	33.2	5.0	2.0	10.0
1-12			<i>Laromer LR 8986</i>	33.2	49.8	5.0	2.0	10.0
1-13			<i>Ebecryl 605</i>	49.8	33.2	5.0	2.0	10.0
1-14			<i>Ebecryl 605</i>	33.2	49.8	5.0	2.0	10.0

[0081] The comparative oligomers have, according to the technical specifications in the brochures of the corresponding companies, the following composition:

Craynor CN 435 (ex Cray Valley; product guide "Photocure and specialty monomers", May 1997): low viscosity polyether acrylate (170 mPa.s);
 Laromer LR 8765 (ex BASF AG; raw materials for radiation-curing coatings EDL 9716) : aliphatic epoxy acrylate (600-1200 mPa.s);
 Laromer LR 8986 (ex BASF AG; Surfex 98, June 1998, Harrogate and MSDS ED 02266-3 dd. 98-07-23) : aromatic epoxy acrylate, viscosity (3000-6000 mPa.s);
 Ebecryl 605 (ex UCB; radiation curable products 8108/02/98) : 75%
 Ebecryl 600 (= bisphenol A epoxy di-acrylate) and 25 % TPGDA (viscosity 7500 mPa.s).

[0082] After measuring viscosities, each ink composition was coated repeatedly on a clear unsubbed 100 µm thick polyester film, using a bar coater and a 10 µm wired bar. The coated films were placed on a conveyer belt and transported underneath a UV lamp. A Fusion DRSE-120 conveyer, equipped with a Fusion VPS/I600 lamp (H bulb), powered at 60% and/or 100% of the nominal maximum input power, was used to cure the coated inks. Different transportation speeds are used for the same coated ink composition in order to determine the maximum speed at which curing is possible. The lowest belt speed that could be used with the conveyer was 9 m/min, the highest was 70 m/min. By means of a scratch test with a cotton bud the curing was visually evaluated: when the coating did not remain visually unchanged after scratching, the curing was not complete. The highest curing speed, was the highest transportation speed at which the coating remained unchanged after scratching. This maximum speed is indicated in table 3, together with the viscosities of the corresponding ink.

TABLE 3

free radical polymerization of ink compositions with the oligomers of the invention		
Ink number	Viscosity (mPasec)	Maximum curing speed (m/min) when Power is 60% of maximum P
1-1	24	70
1-2	18	50
1-3	50	50
1-4	32	50
1-5	30	50
1-6	20	50
1-7	44	30
1-8	26	30
1-9	80	50
1-10	38	50
1-11	122	70
1-12	45	70
1-13	125	50
1-14	46	50

[0083] As one can see from table 3, the inks wherein the oligomer of the invention has been replaced by a comparative oligomer show the following results:

- ink numbers 1-7 and 1-8 of the radiation curable ink show an almost equal viscosity, but curing of these inks resulted in a lower curing speed
- the other ink numbers 1-9 up to 1-14 have a (much) higher viscosity and curing of these inks resulted in the same curing speed as the invention.

[0084] The lower viscosity of the inks of the invention - in combination with high curing speed - is of utmost importance if one wants to use these inks in ink jet printheads.

Example 2:

free radical polymerization of black inks

[0085] The general composition of the black colored radiation curable inks of the invention was:

*Ultraviolet polymerizing amino modified polyetheracrylate of the invention (see above; table 1);
Ultraviolet polymerizing reactive diluent Radcure DPGDA (UCB, dipropylene glycol diacrylate);
N-methyl diethanol amine (NMDA) as coinitiator or synergist; ethanol;
isopropylthioxanthone (Quantacure ITX, Rahn AG);
Solpserse 24000SC (Zeneca) (10 wt% solution in Radcure DPGDA);
Special Black 250 (Degussa).*

[0086] Substituting the oligomer of the invention by another oligomer gave rise to a comparative ink composition.

[0087] All inks were prepared on a basis of a total final weight of 20 g. All inks contained 30.9 wt% of the oligomer, 2% wt of NMDA, 2% wt of ethanol and 10% wt of Quantacure ITX. The ink compositions are represented in table 4.

[0088] Firstly, the Radcure DPGDA was added to the oligomer of the invention or to the comparative oligomer. The

resulting mixture was stirred for a couple of minutes until the added diluent was completely dissolved. As a third ink component 10 wt % admixture of Solsperse 24000SC in DPGDA was added while stirring for about five minutes to complete the solution step. After further adding the Special Black, the resulting ink was milled for 24 hour in a ball mill. Furtheron, the NMDA and the ethanol were added, and the resulting mixture was stirred for about 5 minutes. As the last compound the solid photoinitiator Quantacure ITX was added. The resulting mixture was stirred for about 1 hour in order to completely dissolve the ITX. The ink compositions that were tested are represented in table 4.

TABLE 4

ink composition of black colored inks of the invention for free radical polymerization								
Ink number	oligomer of the invention comparative oligomer		wt% DPGDA	wt% NMDA	wt% ethanol	wt% sol-sperse 24000CS	wt% Special Black 250	wt% Quantacure ITX
	name	wt%						
2-1	Craynor CN501	30.9	46.35	5.0	2.0	0.75	5.0	10.0
2-2	Laromer LR8869	30.9	46.35	5.0	2.0	0.75	5.0	10.0
2-3	Laromer LR8889	30.9	46.35	5.0	2.0	0.75	5.0	10.0
2-4	Craynor CN435	30.9	46.35	5.0	2.0	0.75	5.0	10.0
2-5	Laromer LR8765	30.9	46.35	5.0	2.0	0.75	5.0	10.0
2-6	Laromer LR8986	30.9	46.35	5.0	2.0	0.75	5.0	10.0
2-7	Ebecyl605	30.9	46.35	5.0	2.0	0.75	5.0	10.0

[0089] After the viscosities had been measured, each ink composition underwent a similar treatment as the inks described in example 1: barcoating, curing at different belt speed and determining the maximum curing speed. Results of the experiments are given in table 5.

TABLE 5

free radical polymerization of black ink compositions with the monomers of the invention		
Ink number	Viscosity (mPa.s)	Maximum curing speed (m/min) at 60% power
2-1	51	40
2-2	70	50
2-3	40	40
2-4	62	30
2-5	>100	50
2-6	>100	50
2-7	>100	50

[0090] As can be seen from table 5, all black inks wherein the oligomer of the invention was replaced by another oligomer showed a (much) higher viscosity of the radiation curable ink and a lower curing speed.

[0091] The lower viscosity of the inks of the invention is of utmost importance if one wants to use these inks in ink jet printheads.

Claims

1. A radiation curable ink composition for ink jet printing comprising as radiation polymerizable compound an amino functionalized polyetheracrylate.
2. A radiation curable ink composition according to claim 1 wherein said amino functionalized polyetheracrylate is derived from trimethylolpropanetriacrylate.
3. A radiation curable ink composition according to claim 1 or 2 further comprising a mono-, di-, or polyfunctional reactive diluent.
4. A radiation curable ink composition according to claim 3 wherein said reactive diluent is chosen from the group consisting of acrylates, vinyl ethers, vinyl etheracrylates and vinyl ethermethacrylates.
5. A radiation curable ink composition according to claim 3 or 4 wherein said amino functionalized polyether acrylate is present in an amount of at least 30 % of all polymerizable compounds present.
6. A radiation curable ink composition according to any of claims 1 to 5 further comprising a photoinitiator.
7. A radiation curable ink composition according to claim 6 further comprising an initiator synergist.
8. A radiation curable ink composition according to any of claims 1 to 7 further comprising a colorant.
9. A radiation curable ink composition according to claim 8 wherein said colorant is a dispersed pigment.
10. A radiation curable ink composition according to any of claims 1 to 9 further comprising a solvent in a concentration between 1 and 10 %.
11. A radiation curable ink composition according to any of claims 1 to 10 the viscosity of which is comprised between 1 and 75 mPa.s at 25 °C.
12. A radiation curable ink composition according to any of claims 1 to 11 further comprising a dendrimer.
13. A process for obtaining a monochrome or multicolour ink jet image comprising the steps of jetting one or more streams of ink droplets having a composition according to any of the previous claims to a receiver element, and subjecting the obtained image to radiation curing.
14. A process according to claim 13 wherein said radiation curing is performed by means of one or more ultra-violet sources or electron beam sources.



European Patent
Office

EUROPEAN SEARCH REPORT

Application Number
EP 99 20 2795

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
A	US 5 734 002 A (JAEGER ULRICH ET AL) 31 March 1998 (1998-03-31) * column 2, line 34 - column 4, line 67; examples 7-12 *	1,2,6,11	C09D11/10 C09D11/00 C09D171/02 C08G65/333
A	WO 97 47478 A (THOMAS DE LA RUE INTERNATIONAL ;HOWLAND PAUL (GB); DRINKWATER KENN) 18 December 1997 (1997-12-18) * page 13, line 16 - line 30 *	1,2,6-9	
A	US 3 925 349 A (GASKE JOSEPH E) 9 December 1975 (1975-12-09) * column 1, line 15 - line 39 * * column 6, line 35 - line 68 *	1,2,6	
A	DE 23 49 280 A (BASF AG) 24 April 1975 (1975-04-24) * page 4, paragraph 4 - page 5, paragraph 1; claims 1,2 *	1,2	
			TECHNICAL FIELDS SEARCHED (Int.Cl.7)
			C09D C08G C08F
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 1 February 2000	Examiner Miller, A
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document</p>			

EPO FORM 1503 03.82 (P04C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 99 20 2795

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report.
The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

01-02-2000

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 5734002 A	31-03-1998	DE 19508310 A	12-09-1996
		DE 19508308 A	12-09-1996
		CA 2171264 A	10-09-1996
		EP 0731121 A	11-09-1996
		JP 8259660 A	08-10-1996
		US 5977284 A	02-11-1999
WO 9747478 A	18-12-1997	AU 3100097 A	07-01-1998
		CA 2258251 A	18-12-1997
		DE 19781815 T	17-06-1999
		GB 2330111 A	14-04-1999
US 3925349 A	09-12-1975	US 3844916 A	29-10-1974
		BE 804019 A	17-12-1973
		CA 1016692 A	30-08-1977
		DE 2346424 A	11-04-1974
		FR 2200297 A	19-04-1974
		GB 1448737 A	08-09-1976
		JP 1054956 C	23-07-1981
		JP 49094736 A	09-09-1974
		JP 55000430 B	08-01-1980
		NL 7312860 A,C	20-03-1974
		US 3914165 A	21-10-1975
DE 2349280 A	24-04-1975	NONE	

EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82